

CANDIDATE NUMBER.....



AFRICA
UNIVERSITY
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"Investing in Africa's Future"

COLLEGE OF HEALTH, AGRICULTURE & NATURAL SCIENCES
DEPARTMENT OF HEALTH SCIENCES
BACHELOR OF MEDICAL LABORATORY SCIENCES HONOURS DEGREE

NSLS103: CLINICAL PATHOLOGY

END OF SECOND SEMESTER FINAL EXAMINATIONS

APRIL/MAY 2019

LECTURER: MR G. MALUNGA

DURATION: 3 HOURS

INSTRUCTIONS

Write your candidate number on the space provided on top of each page

Answer **all** questions in sections A on the question paper.

Answer **all** questions in section B on separate answer sheets provided.

Answer any **3** questions in section C on separate answer sheets provided

Credit will be given for logical, systematic and neat presentations in sections B and C

SECTION A : MULTIPLE CHOICE [40MARKS]

- Answer all questions by encircling the correct response T for TRUE or F for FALSE for each statement in all the questions
- Each correct response is allocated half mark

1. The following are functions of the urinary system
T F a) Excretion
T F b) Water balance
T F c) Acid –base balance
T F d) Reproduction
2. If plasma pH is low, rate of
T F a) H^+ excretion increases
T F b) HCO_3^- excretion decreases
T F c) H^+ excretion decreases
T F d) HCO_3^- excretion increases
3. Blood urea levels can be affected by
T F a) Dietary protein levels
T F b) Liver disease
T F c) Renal insufficiency
T F d) Glomerular membrane damage
4. Molecules are reabsorbed from the nephron into the capillary through
T F a) Filtration
T F b) Diffusion
T F c) Active transport
T F d) Osmosis
5. The following hormones are produced in the kidneys
T F a) Rennin
T F b) Erythropoietin
T F c) Vitamin D3
T F d) Aldosterone
6. A urine dipstick detects the following
T F a) Leucocytes
T F b) Blood
T F c) Urine casts
T F d) Nitrite

7.

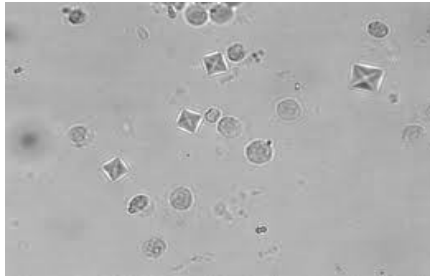


Fig 1

The diagram in Fig 1 shows

- | | | |
|---|---|------------------------------|
| T | F | a) White Blood Cells |
| T | F | b) Calcium Oxalates |
| T | F | c) Granular Casts |
| T | F | d) <i>S. haematobium</i> ova |

8. Electrolytes which are of medical importance are

- | | | |
|---|---|---------------------|
| T | F | a) Na^+ |
| T | F | b) HCO_3^- |
| T | F | c) Al^{3+} |
| T | F | d) Cl^- |

9. The following refers to electrolytes

- | | | |
|---|---|--|
| T | F | a) Na^+ is the major extracellular cation |
| T | F | b) Cl^- exists in equal amounts both in ECF and ICF |
| T | F | c) K^+ is the major intracellular cation |
| T | F | d) HCO_3^- is an extracellular ion |

10. **All** Electrolytes can be measured using

- | | | |
|---|---|---|
| T | F | a) Spectrophotometry |
| T | F | b) Ion Selective Electrodes |
| T | F | c) Coulometric - Amperometric Titration |
| T | F | d) Enzymatic methods |

11. The measurement of the following analyte/s is **greatly** affected by haemolysis

- | | | |
|---|---|---------------------|
| T | F | a) K^+ |
| T | F | b) HCO_3^- |
| T | F | c) Urea |
| T | F | d) Cl^- |

12. The following refers to blood gases

- | | | |
|---|---|---|
| T | F | a) Some of the CO_2 is transported in gaseous state in the blood |
| T | F | b) O_2 is transported bound to haemoglobin in the blood |
| T | F | c) Some of the oxygen is carried dissolved in plasma |
| T | F | d) CO_2 is transported in the form of HCO_3^- |

13. The panel of blood gas analysis include

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- | | | |
|---|---|---------------------|
| T | F | a) HCO_3^- |
| T | F | b) Pco_2 |
| T | F | c) Po_2 |
| T | F | d) pH |
14. Samples for blood gas analysis
- | | | |
|---|---|---|
| T | F | a) Must be collected in blood tubes containing an anticoagulant |
| T | F | b) Can be collected from veins |
| T | F | c) Must be sent to the laboratory on ice |
| T | F | d) Must be centrifuged before analysis |
15. The following laboratory processes are done on a CSF sample
- | | | |
|---|---|----------------------------------|
| T | F | (a) ZN staining |
| T | F | (b) Urea and protein measurement |
| T | F | (c) WBC and RBC counts |
| T | F | (d) Geimsa staining |
16. The biochemical analyte/s which is/are commonly measured in pericardial fluid, ascitic fluid and pleural fluid is/are
- | | | |
|---|---|-------------|
| T | F | a) Chloride |
| T | F | b) Protein |
| T | F | c) LDH |
| T | F | d) Glucose |
17. A transudative body fluid has the following laboratory findings
- | | | |
|---|---|--------------------------|
| T | F | a) High specific gravity |
| T | F | b) High WBC |
| T | F | c) Low RBC |
| T | F | d) A cloudy appearance |
18. Plasma levels of calcium are influenced by the following
- | | | |
|---|---|------------------|
| T | F | a) Renal disease |
| T | F | b) Vitamin D |
| T | F | c) Calcitonin |
| T | F | d) Malabsorption |
19. The following are water-soluble vitamins
- | | | |
|---|---|---------------------------|
| T | F | a) Vitamin B ₆ |
| T | F | b) Vitamin D |
| T | F | c) Vitamin C |
| T | F | d) Vitamin K |

20. The following refers to phosphate in the human body
- | | | |
|---|---|--|
| T | F | a) 85% of it is present in the skeleton |
| T | F | b) 45% of serum phosphate exist as free ions |
| T | F | c) 99% of it is present in the skeleton |
| T | F | d) 40% exist bound to proteins |

SECTION B: [20 MARKS]

Answer all questions on separate answer sheets provided

1. State 5 functions of parathyroid hormone. [5]
2. What are the features of folate deficiency. [5]
3. Calculate the anion gap and explain its significance, for a diabetic patient with the following laboratory results :
 $\text{Na}^+ = 136 \text{ mmol/l}$
 $\text{K}^+ = 5 \text{ mmol/l}$
 $\text{Cl}^- = 97 \text{ mmol/l}$
 $\text{HCO}_3^- = 13 \text{ mmol/l}$ [5]
4. Name any 5 aspirates which can be analyzed in a clinical laboratory. [5]

SECTION C : [75 marks]

Answer any 3 questions from this section on separate answer sheets provided

1. Explain how blood buffers function in maintaining optimal blood pH. [25]
2. Give an account of vitamin deficiency diseases. [25]
3. Describe the laboratory diagnosis of acute renal failure. [25]
4. Describe how a CSF sample is processed in a clinical laboratory. [25]
5. Discuss the regulation of Calcium and Phosphate in the body. [25]